

Inhibition of Cu-Amyloid- β by using Bifunctional Peptides with β -Sheet Breaker and Chelator Moieties

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Although the exact origin of the sporadic form of Alzheimer's disease (AD) is not known, it is clear that this pathology is multifactorial. One of the most prominent factors is the peptide amyloid- β (A β). β - and γ -secretase cleave this peptide from the amyloid precursor protein. In healthy brain tissue, A β exists in a nontoxic, monomeric form; however, AD patients present brain tissue with amyloid plaques, which are insoluble aggregates of peptides with a high concentration of A β 40 and A β 42 (two forms of A β containing 40 and 42 amino acids, respectively). Plaque formation occurs through a commonly accepted progression termed the amyloid cascade hypothesis in which an accumulation of A β monomers leads to the formation of A β oligomers, protofibrils, and finally fibrils. Although A β fibrils are the main components of plaques, recent studies indicate A β oligomers as the most neurotoxic species.^[1,2]

Linked to A β is another factor thought to be important to AD progression: oxidative stress. Redox-active metal ions

such as copper and iron play a key role in formation of reactive-oxygen species (ROS), and evidence has been provided that copper ions bound to A β might be involved in the formation of ROS in AD.^[3-5] Cu-A β catalyzes ROS production under aerobic conditions in vitro in the presence of physiological reductants, such as ascorbate.^[3,6,7]

In this respect, inhibition of the formation of toxic A β aggregates (oligomers and others) and disruption of Cu-A β with subsequent redox-silencing of Cu have been considered promising strategies against AD progression. Several copper-binding ligands have been tested in vitro and the clioquinol derivative PBT2 showed promising results in a clinical phase IIA study.^[8,9] In addition, compounds able to inhibit A β aggregation have been widely studied,^[10] including peptides,^[11] such as KLVFF (A β 16–20).^[12,13] The hydrophobic A β 16–20 region of full-length A β is implicated in both A β -dimer formation, and in subsequent oligomerization and fibrillization. Thus addition of the short peptide A β 16–20 is expected to disturb the interaction between two molecules of A β to form a dimer and hence inhibit aggregation.

Considering the multifactorial nature of AD etiology, a multifaceted approach for AD treatment is appealing.^[14-17] Herein, we present a dual approach to inhibit A β aggregation and ROS production by combining two functions on a single peptide, that is, the copper chelator and β -sheet breaker moieties (Scheme 1). The idea was to use a sequence

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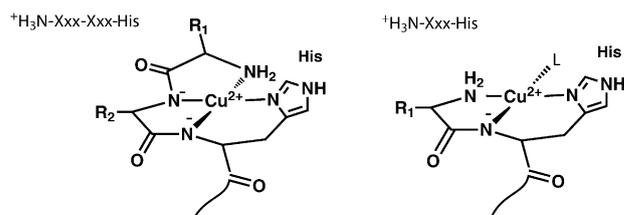
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peptide	Metal-binding function	β -sheet breaker function
KLVFF		¹ H ₃ N-Lys-Leu-Val-Phe-Phe-COO ⁻
A β 12-20		¹ H ₃ N-Val-His-His-Gln-Lys-Leu-Val-Phe-Phe-COO ⁻
A β 13-20		¹ H ₃ N-His-His-Gln-Lys-Leu-Val-Phe-Phe-COO ⁻



Scheme 1.

of the native A β , that is, A β 12–20 and A β 13–20 (Scheme 1), in which the β -sheet breaker sequence KLVFF is elongated by three or four amino acids to include the well-known Cu^{II}-binding sites of the H₂N-Xxx-His and H₂N-Xxx-Xxx-His, respectively.^[18] Peptides with the N-terminal sequence H₂N-Xxx-His and H₂N-Xxx-Xxx-His bind Cu^{II} differently and with a stronger affinity for Cu^{II} than A β and their Cu^{II} complexes do not produce ROS efficiently (Scheme 1).^[19–22] Note that although the peptides A β 12–20 and A β 13–20 contain His13 and 14, which are involved in Cu binding of full length A β , the coordination of Cu^{II} is different due to the presence of an N-terminal amine at position 12 or 13, (which is not the case for A β). The present conceptual work reports on the efficiency of the bifunctional A β 12–20 and A β 13–20 peptides as a β -sheet breaker and a chelator/redox silencer.

Cu²⁺-binding to A β can be followed by the fluorescence of tyrosine (Tyr) at position 10, because tyrosine fluorescence is quenched when A β binds Cu²⁺, and the fluorescence returns after withdrawal of Cu²⁺ from A β . Peptides with the motifs H₂N-Xxx-His and H₂N-Xxx-Xxx-His have been shown to abstract copper from A β .^[20–23] We observed that addition of up to one equivalent of Cu²⁺ reduced the Tyr fluorescence intensity of A β 42 (Figure 1), and subse-

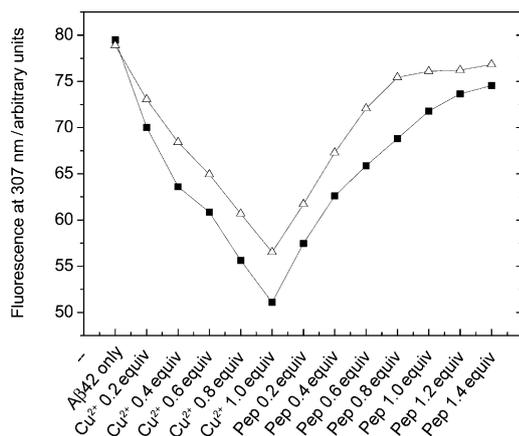


Figure 1. Tyrosine fluorescence from A β 42 (20 μ M) in HEPES buffer (50 mM, pH 7.4) upon the addition of 0–20 μ M Cu²⁺ and 0–28 μ M peptides (Pep) (A β 12–20: ■; A β 13–20: Δ). The fluorescence intensities were calibrated to account for the change in volume. (Excitation at 274 nm, emission at 307 nm).

quent addition of peptide (A β 12–20, A β 13–20) solutions almost entirely reversed the fluorescence quenching at one equivalent of peptide. This suggests that the peptides have a much stronger affinity for copper than A β 42. Similar results were obtained for Cu^{II}-A β 40 and Cu^{II}-A β 16 (not shown).

The effectiveness of bifunctional peptides in diminishing ROS production by Cu-A β 42 was assessed by using the HO[•] trap coumarin-3-carboxylic acid (CCA). The reaction product 7-OH-CCA was measured by fluorescence, which correlates directly with the concentration of HO[•]. The kinetic

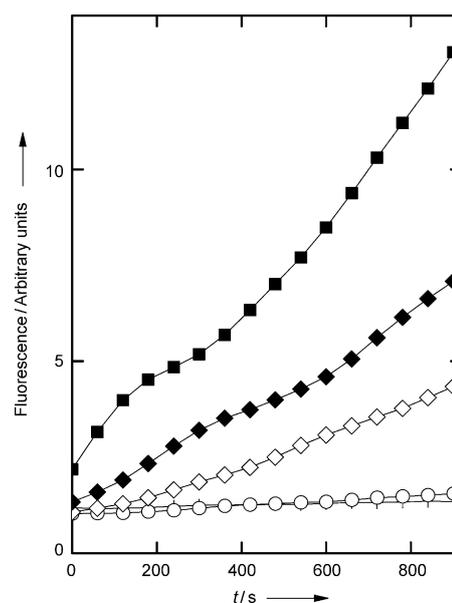


Figure 2. Initial slope of 7-OH-CCA fluorescence intensity for solutions of Cu^{II}-A β 42 (■) and Cu^{II}-A β 42 with either one equivalent of A β 42 (◆), A β 12–20 (+), A β 13–20 (○) or KLVFF (◇). HO[•] production was quantified by measurements of 7-OH-CCA fluorescence for solutions with the conditions: phosphate buffer (6 mM, pH 7.4), desferal (1 μ M), CCA (0.5 mM, pH 9), Cu²⁺ (9 μ M), and ascorbate (0.5 mM), with or without peptides (10 μ M each).

traces of 7-OH-CCA fluorescence are shown in Figure 2. In the presence of ascorbate, Cu^{II}-A β 42 produced HO[•] under aerobic conditions in a time-dependent manner as observed previously.^[22] The two bifunctional peptides A β 12–20 and A β 13–20 were able to suppress the HO[•] production of Cu-A β 42 almost completely (<5%, Figure 2). It seems that A β 12–20 suppresses HO[•] more effectively than A β 13–20, which is in line with the redox inertness of Cu-DAHK compared to Cu-GHK regarding the reduction process.^[19] This is in contrast to KLVFF, which does not contain a copper binding sequence and thus only partially suppressed HO[•] production. This is likely due to trapping of HO[•] by KLVFF (instead of undergoing a reaction with CCA) and not due to the Cu binding of KLVFF, because addition of a second equivalent of A β 42 also reduced the fluorescence of 7-OH-CCA.

The formation of amyloid fibrils can be followed by monitoring the fluorescence emission of Thioflavin T (ThT), a benzothiazole dye, which exhibits increased fluorescence intensity at \approx 486 nm when bound to amyloid fibrils. An increase in ThT fluorescence intensity reports on the formation of amyloid fibrils, and thus increased aggregation (Figure 3). For the aggregation experiments with ThT, the 40-amino acid form A β 40 was used because A β 42 aggregated too fast to measure the effect of the bifunctional peptides on the nucleation phase. The aggregation of apo-A β 40 resulted in a typical sigmoidal trace (Figure 3a), whereas Cu-A β 40 had a higher aggregation rate, as previously reported^[24,25] (Figure 3b). The peptides were added at the beginning of the A β aggregation event to assess their capability

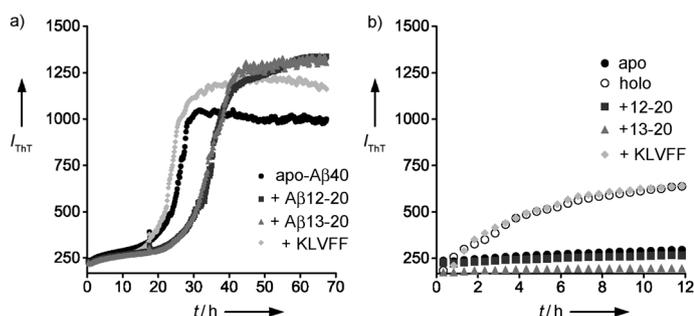


Figure 3. a) Aggregation of apo-A β 40 (40 μ M, black circles) followed by ThT fluorescence with the peptides (40 μ M) A β 12-20 (dark-grey squares), A β 13-20 (mid-grey triangles), or KLVFF (light-grey diamonds) added at the outset. b) Aggregation of A β 40 (40 μ M, black circles) followed by ThT fluorescence with Cu²⁺ (20 μ M, open circles) and peptides (40 μ M) A β 12-20 (dark-grey squares), A β 13-20 (mid-grey triangles), or KLVFF (light-grey diamonds) added at the outset.

to inhibit the formation of amyloid fibrils. Both bifunctional peptides A β 12-20 and A β 13-20 increased the apparent lag-time of A β aggregation (Figure 3a; the retardation of A β aggregation by A β 12-20 and A β 13-20 cannot be due to the binding of residual Cu by A β 12-20 and A β 13-20, because the strong Cu^{II}-binding peptide GHK had no retarding effect (data not shown)). Thus, the bifunctional peptides most likely retard the nucleation step in the aggregation process of apo-A β 40.^[26] Surprisingly, the KLVFF peptide, which only has the beta-sheet breaker functionality, did not inhibit aggregation of A β 40 under these conditions.

In the case of the Cu-induced aggregation of A β 40 the situation was different. First, Cu^{II} binding to A β 40 induced fast aggregation as previously reported.^[25] Both bifunctional peptides inhibited the Cu-induced aggregation of A β 40. This is in line with the capability of the bifunctional peptide to withdraw Cu^{II} from A β 40 (see above). As expected, KLVFF had no effect, which is in agreement with the absence of a Cu-binding domain.

To test the capability of the bifunctional peptides to reverse aggregation, that is, to resolubilize A β 40 aggregates, the bifunctional peptides were added after completed aggregation of apo- and Cu^{II}-A β 40. No significant resolubilization was observed (see the Supporting Information, Figure S1) indicating that these peptides have a retarding effect on aggregation, but are not able to reverse already aggregated amyloids. This suggests that the effect of the bifunctional peptides is primarily kinetic, and that they are interacting on the initially formed oligomers rather than on the end-stage aggregates.

To probe the observed properties of the bifunctional peptides in a more biological environment, we measured the effect of the bifunctional peptides on cell viability in the presence of ascorbate and Cu²⁺ with A β 42. The conditions were similar to the in vitro ROS measurements (see above) and to our previous work.^[13,27] 3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) was used as an indicator of cell toxicity. The survival rate of the treated cells was determined based on a comparison with the absorbance

of untreated cells, which were set to 100% viability (Figure 4). The cell viability was significantly decreased (to about 45%) after addition of Cu-A β 42 in the presence of ascorbate. Addition of A β 12-20 rescued the cells partially, resulting in 66% cell viability. In contrast, A β 13-20 did not show any protecting effect.

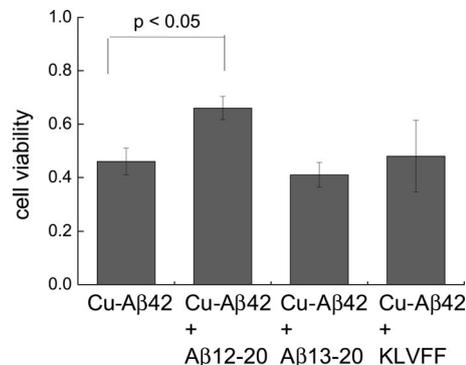


Figure 4. SH-SY5Y cell viability observed based on incubation (24 h) with A β 42 (50 μ M), Cu²⁺ (40 μ M), ascorbate (300 μ M), and peptides (50 μ M): A β 12-20, A β 13-20, and KLVFF. Relative cell viability is given compared to the control.

In conclusion, the two bifunctional peptides used here have demonstrated two designed functions, that is, they were able to withdraw Cu^{II} from A β , inhibiting HO[•] formation and they showed β -sheet breaking activity. Although it is clear that the two bifunctional peptides can exert the two individual functions, further studies are needed to determine whether or not an overall enhanced performance (synergism) is present. Since A β 12-20, and not A β 13-20, showed protecting activity against Cu-A β cell toxicity in the presence of ascorbate in cultured cells, this might support the importance of having a H₂N-Xxx-Xxx-His motif compared with H₂N-Xxx-His. This work shows a proof of concept for bifunctional compounds by using simple and easily synthesized peptides. Peptide A β 12-20 could also be easily modified into a peptidic bifunctional prodrug, which could be activated by β -secretase as has been shown for the H₂N-Xxx-Xxx-His motif.^[28] Further studies are needed to improve and better understand the action of these bifunctional peptides, before attempts for stabilization and vectorization can be envisaged. This is necessary due to the fast degradation of peptides and their low brain penetration, but is expected to be achievable as has been shown for other β -sheet breaker peptides.^[11,13,29]

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